

**Technical Support Bulletin 2005-03** 

**Sample Preservative** 

Aug 30, 2005

Alert Level: ADVISORY

Software Version: All RAS versions

Instrument: All RAS-100 and RAS-500 samplers

Subject: Water Sample Preservative

**Summary:** HgCl<sub>2</sub> is a popular preservative for oceanic water samples. Ultimately, the choice and concentration of preservative for a time-series instrument is a part of science. Thus, the final choice of preservative must be made by the user to fit the scientific objective.

**Potential Negative Effects:** Adding preservative changes the nature of seawater and particulate samples. HgCl<sub>2</sub> preservative is not recommended for salinity measurements with an electrode-based sensor. An inductivity-coupled salinometer can be used under high-HgCl<sub>2</sub> concentration samples (personal communication, Dr. K. Hayashi, MOI, 2003).

**Technical Details:** In order to maintain original nutrients in water samples including NO-x, Si(OH)<sub>4</sub>, PO<sub>3</sub> and DIO, up to 18 months in water temperatures from 0° to 20°C collected by a RAS, McLane recommends a sample solution with approximately 120 ppm HgCl<sub>2</sub>. HgCl<sub>2</sub> solutions with a variety of concentration up to 800 ppm were tested and determined to be no more effective than the original, diluted case. Kattner (1999) reported that sea water nutrients can be preserved as long as 2 years in a 105 HgCl<sub>2</sub> solution. A more diluted solution, 30 ppm, might work for a long-term preservation in an aphotic water column but this concentration has not been tested extensively.

McLane recommends a pre-installation of 5 ml saturated HgCl<sub>2</sub> solution to the sample bag a few days before the RAS deployment. Although solubility of HgCl<sub>2</sub> in distilled water depends on water temperature, 1ml saturated HgCl<sub>2</sub> solution is expected to include 35 mg to 75 mg HgCl<sub>2</sub> at a water temperature between 0° and 20 °C. Under these conditions the HgCl<sub>2</sub> concentrations in 480 ml seawater (approximate volume of a sample bag) is 80 ppm (minimum) to 160 ppm.

Kattner, G., 1999. Storage of dissolved inorganic nutrients in seawater: poisoning with mercuric chloride. Marine Chemistry 67, 61-66.

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